

Jinshan Huang, Yongfei Wu and
Xinde Zhu*Department of Chemistry, Central China Normal
University, Wuhan, Hubei 430079, People's
Republic of China

Correspondence e-mail: tzb@mail.ccnu.edu.cn

Key indicators

Single-crystal X-ray study
 $T = 292\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$
 R factor = 0.052
 wR factor = 0.140
Data-to-parameter ratio = 14.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Bis(μ -2,4-dihydroxyacetophenone salicyl-
hydrazidato)bis[pyridinezinc(II)] dimethyl-
formamide disolvate

The tridentate Schiff base ligand 2,4-dihydroxyacetophenone salicylhydrazide, derived from the condensation of 2,4-dihydroxyacetophenone and salicylhydrazide, in the presence of pyridine and N,N' -dimethylformamide, forms a dinuclear distorted square-pyramidal Zn complex with $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$, viz. $[\text{Zn}_2(\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}_4)_2(\text{C}_5\text{H}_5\text{N})_2] \cdot 2\text{C}_3\text{H}_7\text{NO}$. The complex is centrosymmetric and the two Zn atoms are bridged by two phenolate O atoms.

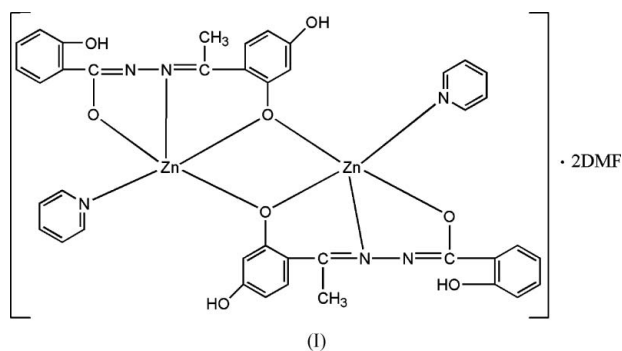
Received 15 June 2005

Accepted 6 July 2005

Online 13 July 2005

Comment

Schiff base compounds have many applications in homogeneous catalysts and antitumor activities (Desai *et al.*, 2001). We report here the synthesis and crystal structure of one such Schiff base complex, namely bis(μ -2,4-dihydroxyacetophenone salicylhydrazidato)bis[pyridinezinc(II)] dimethylformamide disolvate, (I).



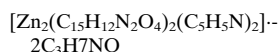
In the crystal structure of (I), there is a centre of symmetry at the mid-point of the $\text{Zn} \cdots \text{Zn}$ vector (Fig. 1). Each Zn atom has a distorted square-pyramidal coordination geometry, formed by one N atom and two O atoms from one Schiff base ligand, one O atom from another Schiff base ligand, and one N atom from a pyridine molecule. Bond lengths and angles involving Zn1 are listed in Table 1.

The crystal structure is stabilized by hydrogen bonds of the types $\text{C}-\text{H} \cdots \text{O}$, $\text{O}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{N}$ (Table 2 and Fig. 2).

Experimental

To a solution of 2,4-dihydroxyacetophenone salicylhydrazide (0.143 g, 0.5 mmol) in DMF (15 ml), $\text{Zn}(\text{OAc})_2 \cdot 2\text{H}_2\text{O}$ (0.110 g, 0.5 mmol) dissolved in pyridine (5 ml) was added with continuous stirring. The mixture was stirred for 5 h at 323 K. After filtration, the yellow solution was allowed to stand at room temperature. Yellow block-shaped crystals were formed by slow evaporation of the solvent at room temperature over a period of 7 d.

Crystal data


 $M_r = 1003.66$

 Monoclinic, $P2_1/c$
 $a = 9.8061 (7) \text{ \AA}$
 $b = 9.9171 (7) \text{ \AA}$
 $c = 23.7427 (17) \text{ \AA}$
 $\beta = 92.545 (2)^\circ$
 $V = 2306.7 (3) \text{ \AA}^3$
 $Z = 2$
 $D_x = 1.445 \text{ Mg m}^{-3}$

 Mo $K\alpha$ radiation

Cell parameters from 2274

reflections

 $\theta = 2.2\text{--}21.6^\circ$
 $\mu = 1.11 \text{ mm}^{-1}$
 $T = 292 (2) \text{ K}$

Block, yellow

 $0.30 \times 0.20 \times 0.20 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer

 φ and ω scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 2001)

 $T_{\min} = 0.733$, $T_{\max} = 0.809$

12351 measured reflections

4511 independent reflections

 3444 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.045$
 $\theta_{\max} = 26.0^\circ$
 $h = -11 \rightarrow 12$
 $k = -11 \rightarrow 12$
 $l = -29 \rightarrow 27$

Refinement

 Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.052$
 $wR(F^2) = 0.140$
 $S = 1.08$

4511 reflections

303 parameters

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0678P)^2]$

 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.50 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.34 \text{ e \AA}^{-3}$

Table 1

 Selected geometric parameters (\AA , $^\circ$).

Zn1—O1	2.014 (2)	Zn1—N3	2.056 (3)
Zn1—O3 ⁱ	2.022 (2)	Zn1—N2	2.071 (3)
Zn1—O3	2.023 (2)		
O1—Zn1—O3 ⁱ	158.56 (11)	O3—Zn1—N3	107.55 (11)
O1—Zn1—O3	103.14 (10)	O1—Zn1—N2	79.19 (11)
O3 ⁱ —Zn1—O3	78.10 (10)	O3 ⁱ —Zn1—N2	86.21 (10)
O1—Zn1—N3	98.03 (11)	O3—Zn1—N2	139.67 (11)
O3 ⁱ —Zn1—N3	101.98 (12)	N3—Zn1—N2	111.98 (12)

 Symmetry code: (i) $-x, -y + 2, -z$.

Table 2

 Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
C16—H16 \cdots O1	0.93	2.39	3.119 (5)	135
O4—H4A \cdots O5 ⁱⁱ	0.82	1.87	2.677 (4)	166
O2—H2A \cdots N1	0.82	1.84	2.561 (4)	146

 Symmetry code: (ii) $x - 1, y, z$.

The methyl H atoms were constrained to an ideal geometry, with C—H distances of 0.96 \AA and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and C—H distances of 0.93 \AA .

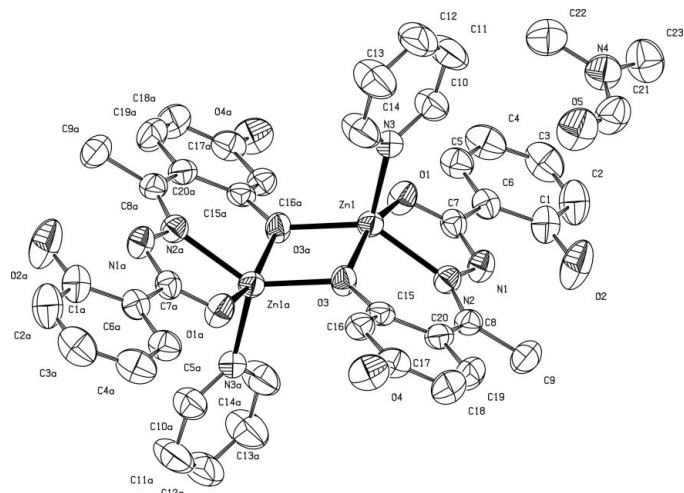


Figure 1

A view of the title complex, showing the labelling of the non-H atoms and 20% probability ellipsoids. The suffix 'a' corresponds to the symmetry position $(-x, 2 - y, -z)$. H atoms have been omitted.

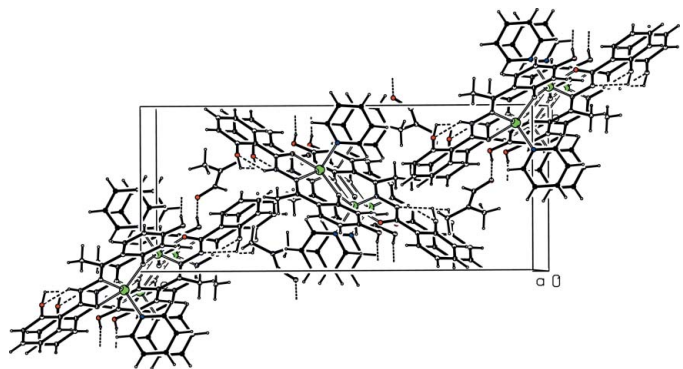


Figure 2

A view of the crystal packing along the a axis. Hydrogen bonds are shown as dashed lines.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT-Plus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 2001); program(s) used to refine structure: SHELXL97 (Sheldrick, 2001); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL.

References

- Bruker (2001). SHELXTL (Version 6.12), SADABS (Version 2.10), SMART and SAINT-Plus. Bruker AXS Inc., Madison, Wisconsin, USA.
 Desai, S. B., Desai, P. B. & Desai, K. R. (2001). *Heterocycl. Commun.* **7**, 83–90.
 Sheldrick, G. M. (2001). SHELXS97 and SHELXL97. University of Göttingen, Germany.